

A New Constructive Method for 1,4-Thiazepine Derivatives<sup>1)</sup>

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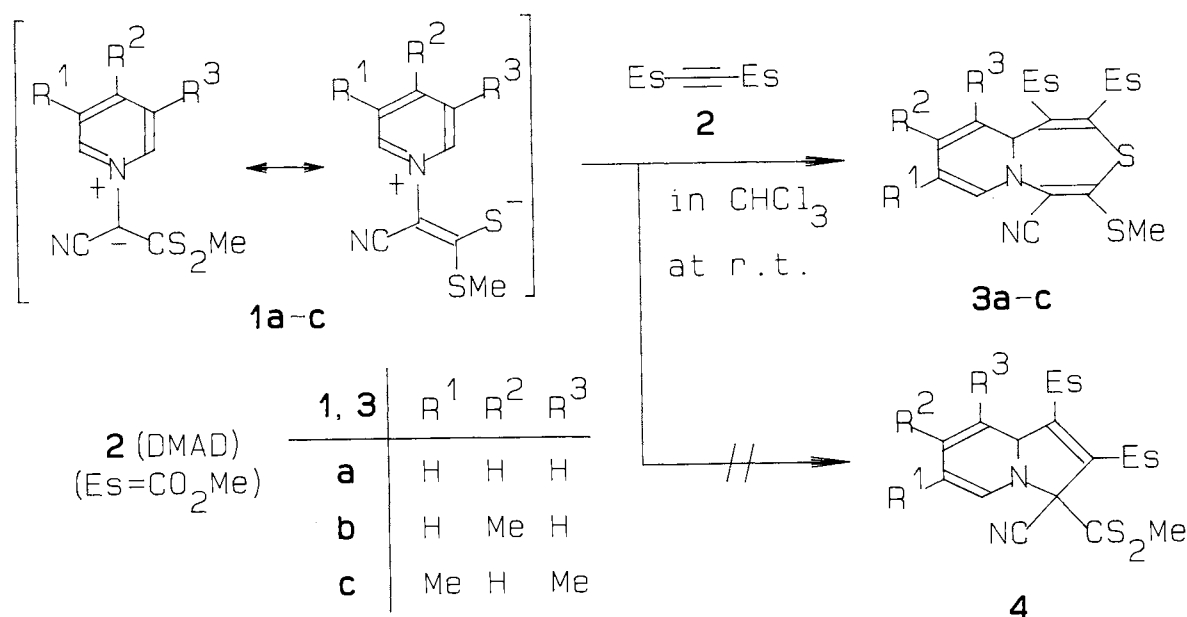
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Some pyridinium 1-[(methylthio)thiocarbonyl]methylides reacted smoothly with dimethyl acetylenedicarboxylate in chloroform at room temperature to afford new heterocycles, 10aH-pyrido[1,2-d][1,4]thiazepine derivatives, in moderate yields.

Recently we have reported a simple and convenient preparative method for indolizines and pyrazolo[1,5-a]pyridines by way of the desulfurization or rearrangement of pyrido[2,1-c][1,4]thiazine and pyrido[1,2-d][1,3,4][1,4]thiazine intermediates.<sup>2)</sup> A characteristic feature in this method is the use of pyridinium salts which can be prepared from the regiospecific S-alkylations of the pyridinium N-methylides or N-aminides substituted by a thiocarbonyl group on the ylidic carbon atom. In continuation of this work, we are interested in the behavior of such pyridinium N-ylides toward electron-poor acetylenic compounds as dipolarophiles, since pyridinium N-ylides, in general, are better known as dipolar species rather than as nucleophiles. Their dipolar cycloadditions and cyclizations were well documented.<sup>3)</sup> In this paper we wish to report the reactions of some pyridinium 1-[(methylthio)thiocarbonyl]methylides with dimethyl acetylenedicarboxylate (DMAD) yielding new nitrogen-bridged heterocycles, dimethyl 2-methylthio-10aH-pyrido[1,2-d][1,4]thiazepine-1,2-dicarboxylates.

When an equimolar mixture of pyridinium 1-[cyano[(methylthio)-thiocarbonyl]]methylide (1a)<sup>2b)</sup> and DMAD (2) was allowed to react in

chloroform at room temperature for 12 h and then the reaction mixture was separated by column chromatography on silica gel using chloroform as an eluent, product **3a**, 34%, mp 127-128 °C,  $\nu$ (KBr) 2218 (CN), and 1725  $\text{cm}^{-1}$  (CO),  $\delta$  ( $\text{CDCl}_3$ ) 2.52 (3H, s, SMe), 3.71 and 3.78 (each 3H, s, COOMe) 5.1-5.7 (2H, m, 8-H and 10-H), 5.87 (br d,  $J=7.0$  Hz, 10a-H), 6.20 (1H, m, 9-H), and 6.65 (1H, br d,  $J=8.0$  Hz, 7-H), was obtained as red needles. Similar treatment of pyridinium N-ylides **1b,c** with the same reagent **2** gave the corresponding red adducts **3b,c** in 30 and 41% yields,<sup>4)</sup> respectively.



The elemental analyses showed that these products were the 1 : 1 adducts between pyridinium N-ylides **1a-c** and DMAD (**2**),<sup>5)</sup> and their <sup>1</sup>H NMR spectra exhibited clearly the presence of a nonaromatic 1,2-dihydropyridine moiety in these molecules because the chemical shifts of the skeletal protons ( $\delta$  5.10-5.65) and methyl protons ( $\delta$  1.71-1.88) were clearly different from those of aromatic indolizines. Of possible structures 1,3-dipolar cycloadduct, 3,8a-dihydroindolizine **4**, was excluded by the indication of an  $\alpha,\beta$ -unsaturated cyano absorption bands (2209-2218  $\text{cm}^{-1}$ ) in the IR spectra. The 10aH-pyrido[1,2-d][1,4]thiazepine structure proposed for **3a-c** was finally confirmed by the single crystal

X-ray structural analysis of **3c**.<sup>6)</sup>

Mechanistically, this reaction seems to proceed via the addition of DMAD (**2**) onto the anionic sulfur atom in ylide **1** followed by the 1,7-cyclization of the resulting zwitterionic species. The driving force of this reaction must be the soft-soft interaction<sup>7)</sup> between the reactive centers of ylide **1** and DMAD (**2**). An alternative path, 1,5-dipolar cycloaddition of **1** with **2**, is less probable because such reaction is unprecedented and its reaction mode ( $\pi 6a + \pi 2s$ ) has to involve a sterically and energetically unfavorable transition state.<sup>8)</sup>

The scope and limitation of this reaction will be described in near future.

#### References

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- 4) **3b**, mp 128-129 °C,  $\nu$  (KBr) 2214 (CN) and 1740  $\text{cm}^{-1}$  (CO),  $\delta$  ( $\text{CDCl}_3$ ) 1.77 (3H, s, 9-Me), 2.49 (3H, s, SMe), 3.68 and 3.75 (each 3H, s, COOMe), 5.11 (1H, br d,  $J=7.0$  Hz, 10-H), 5.11 (1H, br d,  $J=8.0$  Hz, 8-H), 5.79 (1H, br d,  $J=7.0$  Hz, 10a-H) and 6.54 (1H, d,  $J=8.0$  Hz, 7-H). **3c**, mp 129-130 °C,  $\nu$  (KBr) 2209 (CN), 1723 and 1738  $\text{cm}^{-1}$  (CO),  $\delta$  ( $\text{CDCl}_3$ ) 1.77 and 1.88 (each 3H, s, 8-Me and 10-Me), 2.48 (3H, s, SMe),

- 3.69 and 3.77 (each 3H, s, COOMe), 5.75 (2H, br s, 10a-H and 9-H), and 6.23 (1H, br s, 9-H).
- 5) Satisfactory elemental analyses (within 0.3% for C.H.N) were obtained for all new compounds **3a-c**.
- 6) X-Ray crystallography was carried out on a RIGAKU AFC5S diffractometer. The diffraction data were collected with the use of MoK $\alpha$  radiation and 4477 independent reflections were used for solving the structure by the TEXSAN program (TEXSAN TEXRAY, Structure Analysis Package, Molecular Structure Corporation). Crystal data: C<sub>17</sub>H<sub>18</sub>N<sub>2</sub>O<sub>4</sub>S<sub>2</sub>, FW = 378.46, monoclinic, space group P2<sub>1</sub>/n, a = 7.542(2) Å, b = 12.142(2) Å, c = 20.442(2) Å,  $\beta$  = 95.99(2)°, V = 1861.8(5) Å<sup>3</sup>, Z = 4, D<sub>calc</sub> = 1.350 g/cm<sup>3</sup>, R = 0.058, Rw = 0.063. Selected bond lengths (Å) and angles (°): C(1)-C(2) 1.333(7), C(1)-C(10a) 1.533(7), C(2)-S(3) 1.767 (6), S(3)-C(4) 1.765 (6), C(4)-C(5) 1.351 (7), C(5)-N(6) 1.406 (6), N(6)-C(7) 1.405(7), N(6)-C(10a) 1.473(7), C(7)-C(8) 1.327(8), C(8)-C(9) 1.434(8), C(9)-C(10) 1.332(8), C(10a)-C(1)-C(2) 120.3(5), C(1)-C(2)-S(3) 124.2(4), C(2)-S(3)-C(4) 109.1(3), S(3)-C(4)-C(5) 131.7(4), C(4)-C(5)-N(6) 127.3(5), C(5)-N(6)-C(7) 122.4(5), N(6)-C(7)-C(8) 120.1(5), N(6)-C(10a)-C(1) 112.0(4), N(6)-C(10a)-C(10) 109.2(4), C(5)-N(6)-C(10a) 117.6(4), C(7)-N(6)-C(10a) 119.2(4), C(7)-C(8)-C(9) 118.6(6), C(8)-C(9)-C(10) 122.5(6), C(9)-C(10)-C(10a) 118.9(5).
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